## Psymberin, A Potent Sponge-Derived Cytotoxin from *Psammocinia* Distantly Related to the Pederin Family

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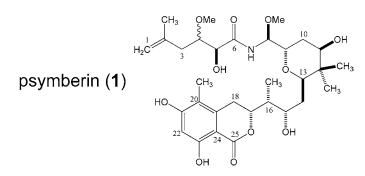
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## ABSTRACT



Bioassay-guided fractionation of the sponge *Psammocinia* sp. afforded psymberin (1) possessing 5*S*,8*S*,9*S*,11*R*,13*R*,15*S*,16*R*,17*R* stereochemistry. Psymberin exhibits structural similarities to the pederin family metabolites. The potent cytotoxicty and unique structural features of 1 make it a promising lead for therapeutic development.

Until now, our efforts to isolate and elucidate the structure of an extremely potent cytotoxin from an undescribed and inconspicuous sponge, *Psammocinia* sp. (Dictyoceratida, Irciniidae), were unsuccessful. For over a decade, we have repeatedly observed that *Psammocinia* extracts collected from the waters of Papua New Guinea exhibited significant solidtumor cytotoxicity. In the course of these studies, a moderately cytotoxic halogenated hexapeptide, cyclocinamide A,<sup>1</sup> and actin inhibitor, swinholide A,<sup>2</sup> were isolated, but their biological activities did not account for the potency of the

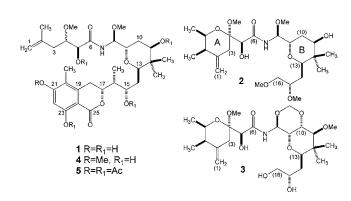
(2) (a) Swinholide A was identified by comparison to data in Kobayashi, M.; Tanaka, J.; Katori, T.; Matsuura, M.; Kitagawa, I. *Tetrahedron Lett.* **1989**, *30*, 2963–2966. (b) Variabilin was identified by comparison to data in Faulkner, D. J. *Tetrahedron Lett.* **1973**, *39*, 3821–3822. (c) Polybrominated phenol ethers were identified by comparison to data in Bowden, B. F.; Towerzey, L.; Junk, P. C. *Aust. J. Chem.* **2000**, *53*, 299–301. Salva, J.; Faulkner, D. J. *Nat. Prod.* **1990**, *53*, 757–760. crude sponge extracts. Furanosesterpenes such as variabilin and polybrominated phenol ethers have also been identified from *Psammocinia*, but these compounds lack significant cytotoxicity.<sup>2</sup> A new approach to this project was adopted with the goal of obtaining the minor constituent possessing the potent cytotoxicity. The revised plan involved a focused bioassay-guided effort utilizing all of the previously prepared fractions from our extract library. Described here are the results of this unusual approach that led to the isolation and study of psymberin (1) (Figure 1), an extremely potent cytotoxin.

Approximately 600 *Psammocinia* fractions prepared between 1990 and 2001 were combined. The methanol-soluble material was subjected to bioassay-guided fractionation against a human colon cancer cell line (HCT-116), yielding  $1^3$  (9.16 × 10<sup>-5</sup>% wet weight of sponge), which exhibited a perplexing array of positive FABMS (NBA matrix) ions including those at *m*/*z* 632 [M + Na]<sup>+</sup>, 610 [M + H]<sup>+</sup>, 578 [M - MeOH + H]<sup>+</sup> (base peak), and 560 [M - H<sub>2</sub>O -

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<sup>(1) (</sup>a) Isolation: Clark, W. D.; Corbett, T.; Valeriote, F.; Crews, P. J. Am. Chem. Soc. **1997**, 119, 9285–9286. (b) Synthesis of a diastereomer: Grieco, P. A.; Reilly, M. Tetrahedron Lett. **1998**, 39, 8925–8928.



**Figure 1.** Structures of psymberin (1), its methylester (4) and acetyl (5) derivatives, pederin (2), and mycalamide A (3) (atom positions analogous to 1 are shown in parentheses).

MeOH + H]<sup>+</sup>. The molecular ion was confidently assigned on the basis of ESI-MS/MS analysis of the m/z 610 ion that fragmented to give m/z 578 and 560 and a negative ESIMS ion that was observed at m/z 608 [M - H]<sup>-</sup>.

The molecular formula for psymberin (1) of  $C_{31}H_{47}NO_{11}$ was consistent with the <sup>13</sup>C NMR DEPT estimate of C<sub>31</sub>H<sub>41</sub> (7 CH<sub>3</sub>, 5 CH<sub>2</sub>, 10 CH, and 9 C) with six additional protons attached to heteroatoms. The obvious functionalities included 2 carbonyl groups, 2 aryl oxygen sites, and 10 sp<sup>3</sup> carbon centers with attached oxygens. Comparison of this oxygen count with the MS-derived molecular formula meant that there were at least three ether groups, two of which were OCH<sub>3</sub>'s. Dereplication based on the <sup>1</sup>H and <sup>13</sup>C NMR data revealed 1 shared several structural features with the pederin family of terrestrial and marine metabolites, as shown in Figure 1 and further discussed below. The geminal protons  $(\delta_{\rm H} 4.69 \text{ and } 4.71)$  of a low field methylene  $(\delta_{\rm C} 111.5)$  were chosen as the starting point for the structure elucidation of substructure A. The  ${}^{2-3}J_{H,C}$  HMBC correlations linked this vinylic methylene to other adjacent carbons, and additional connectivities were established by COSY data between mutually coupled vicinal protons. A  ${}^{2}J_{H,C}$  HMBC correlation between H-5 and an amide carbonyl ( $\delta_{\rm C}$  174.6) established the C-5 to C-6 linkage, while a  ${}^{3}J_{H,C}$  HMBC correlation from

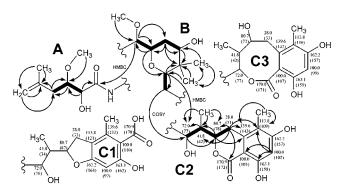


Figure 2. Psymberin (1) partial structures with significant  ${}^{2-3}J_{H,C}$  HMBC and COSY correlations represented as arrows and bold bonds, respectively. Numbers for C1–C3 are experimental and calculated (in parentheses)  ${}^{13}$ C NMR shifts.

 $\delta_{\rm H}$  3.18 to C-4 ( $\delta_{\rm C}$  81.2) permitted the assignment of the attached methoxy. Next, substructure **B** was established and confirmed by 2D NMR experiments. The HMBC correlations shown in Figure 2, especially that from H-8 to C-6 allowed joining substructures **A** and **B**. Thus, psymberin (1) possessed the oxane ring B present in 2 and 3, but unlike all previously reported pederin biomolecules it lacked the signature ring A.

The most challenging aspect of the structure elucidation pertained to defining substructure C, comprising  $C_{13}H_{15}O_5$ . The resonance for H-16 ( $\delta_{\rm H}$  1.87, ddd) served as the anchor point for defining the chain C-15 to C-18 via the COSY correlations shown in Figure 2. Unexpectedly, unraveling the interrelationships of the remaining atoms  $(C_8H_6O_4)$  that were a part of a pentasubstituted aromatic ring proved to be challenging. These hydrogens appeared in the <sup>1</sup>H NMR spectrum as a methyl singlet ( $\delta_{\rm H}$  2.08, CH<sub>3</sub>-16), two exchangeable protons ( $\delta_{\rm H}$  8.58 s and 11.20 s in dioxane- $d_8$ ), and a low field aryl ring singlet ( $\delta_{\rm H}$  6.22, H-21). Initially the remaining nonaromatic ring carbon ( $\delta_{\rm C}$  170.9) was assumed to be a carboxylic acid based on the smooth conversion of 1 to a methylated product 4 (ESIMS m/z 592  $[M - MeOH + H]^+$  (base peak) and 646  $[M + Na]^+$ ).<sup>4</sup> The <sup>1</sup>H NMR spectrum of **4** in dioxane- $d_8$  revealed the persistence of the most downfield exchangeable singlet whose extreme low-field shift was ultimately rationalized as due to intramolecular hydrogen bonding of the phenol proton to a carbonyl. The carbon chemical shifts and significant  ${}^{2-3}J_{H,C}$  HMBC correlations from the aryl proton and CH<sub>3</sub> were initially concluded to be consistent with partial structure C1. Alternatively, structure C2, not initially considered by us but embedded in the structure proposed for irciniastatin A,<sup>5</sup> was also consistent with these data. Another substructural variant, C3, was also in general agreement with the preceding data.

<sup>(3)</sup> Psymberin (1): clear glasslike solid;  $[\alpha]^{27}_{D} = +29$  (*c* 0.02, MeOH); CD (EtOH) 280 ( $\Delta\epsilon$  14), 310 ( $\Delta\epsilon$  -6) nm; IR (film)  $\nu_{max}$  3378, 2935, 1653, 1617, 1508, 1450, 1377, 1252, 1173, 1108, 1064, 970, 898 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD)  $\delta$  6.22 (H-22, s, 1H), 5.36 (H-8, d, J = 8.0 Hz, 1H), 4.71 (1-H<sub>E</sub>, s, 1H), 4.69 (1-H<sub>Z</sub>, s, 1H), 4.46 (H-17, ddd, J = 3.0, 6.0,12.0 Hz, 1H), 4.33 (H-5, d, J = 2.5 Hz, 1H), 3.91 (H-9, ddd, J = 3.0, 6.0,8.0 Hz, 1H), 3.91 (H-15, m, 1H), 3.65 (H-4, ddd, J = 2.5, 3.5, 9.5 Hz, 1H), 3.57 (H-11, dd, J = 4.5, 11.0 z, 1H), 3.47 (H-13, dd, J = 1.0, 11.0 Hz, 1H), 3.33 (8-OCH<sub>3</sub>, s, 3H), 3.18 (4-OCH<sub>3</sub>, s, 3H), 3.11 (H-18<sub>β</sub>, dd, J = 3.0, 17.0 Hz, 1H), 2.83 (H-18 $_{\alpha}$ , dd, J = 12.0, 17.0 Hz, 1H), 2.32 (3-H<sub>b</sub>, dd, J = 9.5, 14.5 Hz, 1H), 2.08 (20-CH<sub>3</sub>, s, 3H), 2.06 (H-3<sub>a</sub>, dd, J = 3.5, 14.5 Hz, 1H), 1.98 (H-10 $_{e}$ , ddd, J = 2.5, 4.5, 13.5 Hz, 1H), 1.87 (H-16, ddq, J = 2.0, 6.0, 6.0, 1H), 1.75 (H-10 $_{a}$ , ddd, J = 6.0, 11.0, 13.5 Hz, 1H), 1.75 (H-14, m, 2H), 1.69 (1-CH<sub>3</sub>, s, 3H), 1.07 (16-CH<sub>3</sub>, d, J = 6.5 Hz, 3H), 0.95 (12-CH<sub>3e</sub>, s, 3H), 0.87 (12-CH<sub>3a</sub>, s, 3H); <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>OD) δ 174.6 (C-6, s), 170.9 (C-25, s), 163.1 (C-23, s), 162.2 (C-21, s), 142.4 (C-2, s), 139.6 (C-19, s), 113.8 (C-20, s), 111.5 (1-CH<sub>2</sub>, t), 99.98 (C-22, d), 99.92 (C-24, s), 81.2 (C-4, d), 80.7 (C-17, d), 80.5 (C-13, d), 78.4 (C-8, d), 72.6 (C-9, d), 72.0 (C-15, d), 71.9 (C-5, d), 70.6 (C-11, d), 56.2 (4-OCH<sub>3</sub>, q), 55.0 (8-OCH<sub>3</sub>, q), 41.8 (C-16, d), 38.3 (C-12, s), 37.2 (C-3, t), 32.9 (C-14, t), 29.0 (C-10, t), 28.0 (C-18, t), 22.2 (12-CH<sub>3e</sub>, q), 21.4 (2-CH<sub>3</sub>, s), 12.6 (12-CH<sub>3a</sub>, q), 9.3 (20-CH<sub>3</sub>, q), 7.7 (16-CH<sub>3</sub>, q).

<sup>(4)</sup> Methylation of **1** was performed as described in Hashimoto, N.; Aoyama, T.; Shioiri, T. *Chem. Pharm. Bull.* **1981**, *29*, 1475–1478. Although we originally favored **C1** as a possible substructure, we are grateful to an anonymous reviewer who noted the potential of phenols to undergo *O*-methylation (Toyohiko, A.; Terasawa, S.; Sudo, K.; Shioiri, T. *Chem. Pharm. Bull.* **1984**, *32*, 3759–3760.).

**Table 1.** Relative Stereochemistry of Subunits 1-3 Showing Selected NOE Enhancements (arrows) and  ${}^{3}J_{H,H}$  Spin Coupling Constants Used for the Configuration Analysis of Psymberin (1) with Comparative Data for Pederin (2)

subunit 1				subunit 2						subunit 3					
configu	configuration of C-4 to C-5 configura				onfiguratio	tion of C-8 to C-13				configuration of C-15 to C-17					
H <sub>4</sub> 0 -2 -2 -2 -2 -2 -2 -2 -2 -2 -2	MeO H3 H0 H0 MeO OH HO 2J (H-5/C-4) =	C5 H5 H0 C3 4R, 5S rotar C6 C3 C3 C3 C3 C3 C3 C3 C3 C3 C3 C3 C3 C3	C5 H4 mers C6 H4 C5 H4 H5 Me	H3C C	HH, 1000 H1000 H1000	H <sub>112</sub> H <sub>112</sub> 13F 0 11R <sup>*</sup> OH	CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	( cs)			H15 OF	6R* 17R*	CH <sub>3</sub> CH <sub>3</sub> C14 <sup>-</sup> H <sub>18α</sub> H <sub>18α</sub> C15 <sup>-</sup>	(C15) ← H15 C H16 C-15/C-161 H16	14 CH <sub>3</sub> rotamers H16 18 C16 CH <sub>3</sub> H17 CH <sub>3</sub> CH <sub>3</sub>
position	4	5	8	9	10 <sub>e</sub>	10 <sub>a</sub>	11	12	13	14	15	16	17	$18_{\alpha}$	$18_{\beta}$
<sup>1</sup> H-NMR data for psymberin (1) <sup><i>a</i></sup>	3.65, ddd, J=2.5, 3.5, 9.5	4.33, d, <i>J</i> =2.5	5.36, d, J=8.0	3.91, ddd, <i>J</i> =3.0, 6.0, 8.0	1.98, ddd, <i>J</i> =2.5, 4.5, 13.5	1.75, ddd, J=6.0, 11.0 13.5	3.57, dd, <i>J</i> =4.5, 11.0	0.95, s (12-CH <sub>3e</sub> ); 0.87, s (12-CH <sub>3a</sub> )	3.47, dd, <i>J</i> =1.0,10.0	1.75, m <sup>b</sup>	3.91, m	1.87, ddq, J=2.0, 6.0, 6.0	4.46, ddd, <i>J</i> =3.0, 6.0, 12.0	2.83, dd, <i>J</i> =12.0, 17.0	3.11, dd. <i>J</i> =3.0, 17
<sup>1</sup> H-NMR data for pederin (2) <sup><i>c</i></sup>	NA <sup>d</sup>	4.31, dd, <i>J</i> =2.2, 2.2	5.39, dd, <i>J</i> =8.0, 9.7 <sup>e</sup>	3.81, ddd, J=2.5, 6.3, 8.0	2.05, ddd, <i>J</i> =2.5, 4.6, 13.4	1.76, ddd, <i>J</i> =6.2, 10.9, 13.4	3.65, dd, <i>J</i> =4.6, 10.9	0.95, s (12-CH <sub>3e</sub> ); 0.88, s (12-CH <sub>3a</sub> )	3.25, dd, <i>J</i> =2.0, 10.2	1.73, ddd; 1.61, ddd <sup>f</sup>	3.3-3.5, m	NA	NA	NA	NA
absolute configuration of pederin (2)	NA	\$	S	S	NA	NA	R	NA	R	NA	S	NA	NA	NA	NA

<sup>*a*</sup> Data recorded in CD<sub>3</sub>OD. <sup>*b*</sup> Isochronouse diastereotopic C-14 protons. <sup>*c*</sup> In CDCl<sub>3</sub>, reported by Perry, N. B.; Blunt, J. W.; Munro, M. H. G.; Thompson, A. M. J. Org. Chem. **1990**, 55, 223–227. See Figure 1 for numbering. <sup>*d*</sup> Not applicable <sup>*e*</sup> Splitting (dd) due to coupling with amide protons and H-9. <sup>*f*</sup> Analogous C-14 protons in mycalamide A (**3**) appear as an overlapping multiplet in CDCl<sub>3</sub> (refer to reference in footnote c).

Distinguishing among substructures C1-C3 involved an evaluation of the experimental versus calculated <sup>13</sup>C shifts. Overall, C2 provided the best agreement with the calculated values (Figure 2). Alternatively there was substantive divergence for many experimental shifts of substructure C1 both in the vicinity of the carboxylic acid ( $\delta_{\rm C}$  100.0, 113.8, 139.6) and beyond ( $\delta_{\rm C}$  41.8, 80.7). The agreement between experimental and calculated data for substructure C3 was good except at two carbons in the lactone ring (at  $\delta_{\rm C}$  72.0, 80.7). The IR data of 1 revealed characteristic C=O stretching bands for an amide (1653 cm<sup>-1</sup>) and  $\alpha,\beta$ unsaturated lactone (1617 cm<sup>-1</sup>) that were unchanged in 4, and the broad band at 3378 cm<sup>-1</sup> was characteristic of a hydrogen-bonded phenol and not a carboxylic acid, thereby eliminating C1. Acetylation of 1 (1:1 acetic anhydride/ pyridine) yielded penta-acetate 5 supporting the presence of five hydroxyl moieties. The slight shift in the H-17 resonance of 1 ( $\delta_{\rm H}$  4.46) versus that of 5 ( $\delta_{\rm H}$  4.42) provided the final evidence to rule out substructure C3.

With the planar structure identified, attention shifted toward defining the relative stereochemistry of **1**. Efforts to crystallize **1** and its acetylated and benzyl ester products proved unsuccessful. Instead, NMR-based methods including NOE enhancements in addition to  ${}^{3}J_{\rm H,H}$  and  ${}^{2-3}J_{\rm C,H}$  coupling constants (HSQMBC) were utilized for the configuration analysis of **1** as outlined in Table 1. A small  ${}^{3}J_{\rm H,H}$  (J = 2.5 Hz) H-4/H-5 *gauche* coupling combined with intermediate  ${}^{2}J_{\rm C,H}$  (J = 3 Hz) couplings between H-5/C-4 and H-4/C-5

indicated the presence of two C-4/C-5 rotamers<sup>6</sup> in subunit 1. Difference NOE data supported this conclusion with three key enhancements from H-5 to H-4, 4-OCH<sub>3</sub>, and H-3<sub>a</sub>. As a result of rotational freedom, these data cannot distinguish between  $4R^*$ ,5S\* and  $4S^*$ ,5S\* configurations. Similarly, two C-4/C-5 rotamers were previously reported for **3**.<sup>7</sup>

In view of the intervening amide functionality, the relative stereochemistry of subunit 2 was examined separately. A chair conformation was deduced for the oxane ring on the basis of diagnostic vicinal spin-spin couplings and was supported by NOE enhancements (H-10<sub>e</sub>/H-11<sub>a</sub>, H-11<sub>a</sub>/12-CH<sub>3e</sub>, and H-11<sub>a</sub>/H-13<sub>a</sub>). Additional NOE enhancements between H-8/H-11<sub>a</sub>, H-8/H-13<sub>a</sub>, and 8-OCH<sub>3</sub>/H-10<sub>e</sub> were corroborated by *gauche* H-8/C-10 (J = 3 Hz) and H-8/C-9 (J = 6 Hz) couplings, confirming the 8*S*\*,9*S*\*,11*R*\*,13*R*\* configuration in compound **1**.<sup>8</sup>

In subunit 3, H-17 showed gauche (J = 3.0 Hz) and anti (J = 12.0 Hz) couplings with H-18<sub>a</sub> and H-18<sub>b</sub>, respectively. In turn, H-17 exhibited an intermediate coupling (J = 6.0 Hz) with H-16 that combined with the observed NOEs suggested the two C-16/C-17 rotamers shown in Table 1. Likewise, the gauche coupling of H-15/H-16 (J = 2.0 Hz) was accompanied by NOE enhancements that supported two C-15/C-16 rotamers. Combined, these data justify a relative stereochemistry of  $15S^*, 16R^*, 17R^*$  in **1**.

Attempts to define the absolute stereochemistry of **1** through the preparation of chiral derivatives were thwarted by the formation of multiple degradation products. Instead the well-defined Cotton effect at the  $n \rightarrow \pi^*$  (ca. 270 nm)

<sup>(5)</sup> Shortly after the completion of this work we learned that iciniastatin A, possessing properties identical to those of psymberin (1) had been disclosed: Pettit, G. R.; Xu, J.-P.; Chapuis, J.-C.; Pettit, R. K.; Tackett, L. P.; Doubek, D. L.; Hooper, J. N. A.; Schmidt, J. M. *J. Med. Chem.* **2004**, *47*, 1149–1152. The name has been retained because of its prior incorporation in the research at collaborating institutions.

<sup>(6)</sup> Matsumori, N.; Kaneno, D.; Murata, M.; Nakamura, H.; Tachibana, K. J. Org. Chem. **1999**, 64, 866–876.

<sup>(7)</sup> Perry, N. B.; Blunt, J. W.; Munro, M. H. G.; Thompson, A. M. J. Org. Chem. **1990**, 55, 223–227.

<sup>(8)</sup> A different  $7R^*$  (same as position C-8 in psymberin, 1) configuration was proposed for irciniastatin A (see ref 5).

transition for a chiral dihydroisocoumarin moiety was employed.<sup>9</sup> The positive Cotton effect at 280 nm clearly justified assignment of a 17-R configuration.

Compound 1 shares certain features with the sponge- and beetle-derived pederin family of metabolites (Table 1).<sup>10</sup> The structures of several pederin biomolecules, such as 2 and 3, have been confirmed by synthesis and X-ray crystallography.<sup>11</sup> These data served as a benchmark to further examine the absolute stereochemistry of 1. The important considerations included: (1) a probable shared biogenetic origin among 1-3, (2) analogous NMR data (Table 1), (3) comparable cytotoxicity,<sup>12</sup> and (4) similar [ $\alpha$ ] value.<sup>13</sup> Therefore, we propose the stereochemistry of 1 as 5*S*,8*S*,9*S*,-11*R*,13*R*,15*S*,16*R*,17*R*.

Thirty-four pederin-like molecules have been described to date (Table S1) from across the globe (Figure S7), and **1** stands apart from these in light of its unique structural features. The mixed PKS/NRPS gene cluster described from an unculturable *Paederus* beetle bacterial symbiont sheds light on the biosynthesis of **2**.<sup>14</sup> Thus, **1** is probably biosynthesized by a similar route as **2**; however, it is the first analogue lacking the A-ring oxane and is also devoid of the 1,3-dioxane ring common among sponge-derived analogues such as **3**. The biosynthetic findings for **2** suggest production of **1** by a sponge microsymbiont and are the basis for its proposed name, psymberin.<sup>15</sup>

Psymberin (1) was initially detected via a disk diffusion assay in which it exhibited phenomenal activity against a human colon cancer cell line (HCT-116) (600 zone units, 900 pg/disk).<sup>16</sup> Interestingly, the potencies of the methylester (4) and penta-acetate (5) derivatives of 1 were significantly abated, exhibiting 17- and 1044-fold reductions in activity, respectively. Compound 1 was further evaluated in vitro

(12) SAR studies have revealed that a 55,85 absolute configuration is required for the potent cytotoxicity observed for analogous pederin-like compounds (see Figure 1 for numbering and ref 10).

(13) (a) All reported members of the pederin family possess optical rotations ranging between  $[\alpha] = +22$  (onnamide F: Vuong, D.; Capon, R. J.; Lacey, E.; Gill, J. H.; Heiland, K.; Friedel, T. *J. Nat. Prod.* **2001**, 64, 640–642) and  $[\alpha] = +172$  (theopederin C: Fusetani, N.; Sugawara, T.; Matsunaga, S. *J. Org. Chem.* **1992**, *57*, 3828–3832). (b) See Table S1 for the optical rotations for all pederin metabolites.

(14) (a) Piel, J. Proc. Natl. Acad. Sci. U.S.A. 2002, 99, 14002–14007.
(b) Piel, J.; Wen, G.; Platzer, M.; Hui, D. ChemBioChem 2004, 5, 93–98.
(15) The name psymberin is derived from Psammocinia + symbiont +

(15) The name psymoorn is derived from *Fsammochila* + symoorn + ped*erin* in view of its relationship to the pederin family.

(16) Bioassay procedures are described in Valeriote, F.; Grieshaber, C. K.; Media, J.; Pietraszkewicz, H.; Hoffmann, J.; Pan, M.; McLaughlin, S. *J. Exp. Ther. Oncol.* **2002**, *2*, 228–236.

**Table 2.** Differential Sensitivities  $(LC_{50})$  of Various Cell Lines to Psymberin (1) as Identified in the NCI Developmental Therapeutics in Vitro Screening Program

cell line	LC <sub>50</sub> (M)	cell line	LC <sub>50</sub> (M)
leukemia		melanoma	
CCRF-CEM	$>\!2.5 imes10^{-5}$	LOX IMVI	$> 2.5  imes 10^{-5}$
HL-60(TB)	$>2.5 imes10^{-5}$	MALME-3M	$^{<}2.5 imes10^{-9}$
K-562	$>2.5 imes10^{-5}$	SK-MEL-2	$>2.5 imes10^{-5}$
MOLT-4	$> 2.5  imes 10^{-5}$	SK-MEL-5	$^{<}2.5 imes10^{-9}$
RPMI-8226	$>2.5 imes10^{-5}$	SK-MEL-28	$1.41 imes10^{-5}$
SR	$>2.5 imes10^{-5}$	UACC-257	$>2.5 imes10^{-5}$
		UACC-62	${<}2.5 imes10^{-9}$
breast cancer		colon cancer	
MCF7	$> 2.5  imes 10^{-5}$	HCC-2998	$3.76 imes10^{-7}$
HS 578T	$>2.5 imes10^{-5}$	HCT-116	${<}2.5 imes10^{-9}$
MDA-MB-435	$^{<}2.5 imes10^{-9}$	HT29	$>2.5 imes10^{-5}$
NCI/ADR-RES	$1.9 imes10^{-5}$	SW-620	$> 2.5  imes 10^{-5}$
T-47D	$1.36 imes10^{-5}$		

against an extended panel of 60 human cancer cell lines. The results of this analysis are very intriguing in light of the selective activity exhibited by psymberin against numerous tumor types (Table 2). For example, several melanoma, breast, and colon cancer cell lines demonstrated considerable sensitivity ( $LC_{50} < 2.5 \times 10^{-9}$  M) to psymberin, and all six of the leukemia cell lines proved comparably insensitive ( $LC_{50} > 2.5 \times 10^{-5}$  M). These data, supporting a >10<sup>4</sup> fold activity differential for **1**, are exceptional in view of the equipotent toxicity observed for other pederin analogues.<sup>17</sup>

The discovery of **1** sheds further light on the structure– activity relationships within the pederin family. Our data indicate the A ring is not essential for activity, while the *N*-acyl aminal moiety is crucial for cytotoxicity.<sup>5</sup> Moreover, the dihydroisocoumarin moiety and/or vinylic methyl termini in **1** may be essential for the differential cytoxicity profile of psymberin. Further tests are underway to determine the in vivo antitumor activity of **1**.

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**Supporting Information Available:** NMR spectra, isolation scheme and modeling data for **1**; tables and figures summarizing the structures, distribution, and sources of pederin metabolites. This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(9) (</sup>a) Krohn, K.; Bahramsari, R.; Florke, U.; Ludewig, K.; Kliche-Spory, C.; Michel, A.; Aust, H.-J.; Draeger, S.; Schulz, B.; Antus, S. *Phytochemistry* **1997**, *45*, 313–320, (b) Arakawa, H. *Bull. Chem. Soc. Jpn.* **1968**, *41*, 2541.

<sup>(10) (</sup>a) Narquizian, R.; Kocienski, P. J. In *The Role of Natural Products in Drug Discovery*; Mulzer, J., Bohlmann, R., Eds.; Ernst Schering Research Foundation Workshop 32; Springer: New York, 2000; pp 25–56. (b) See also Table S1.

<sup>(11) (</sup>a) Furusaki, A.; Watanabé, T.; Matsumoto, T.; Yanagiya, M. *Tetrahedron Lett.* **1968**, *60*, 6301–6304. (b) Kocienski, P.; Narquizian, R.; Raubo, P.; Smith, C.; Farrugia, L. J.; Muir, K.; Boyle, F. T. *J. Chem. Soc., Perkin Trans.* 1 **2000**, 2357–2384. (c) See also ref 10.

<sup>(17) (</sup>a) Burres, N. S.; Clement, J. J. Cancer Res. 1989, 49, 2935–2940.
(b) Kobayashi, J.; Itagaki, F.; Shigemori, H.; Sasaki, T. J. Nat. Prod. 1993, 56, 976–981. (c) West, L. M.; Northcote, P. T.; Hood, K. A.; Miller, J. H.; Page, M. J. J. Nat. Prod. 2000, 63, 707–709. (d) Hood, K. A.; West, L. M.; Northcote, P. T.; Berridge, M. V.; Miller, J. H. Apoptosis 2001, 6, 207–219. (e) Paul, G. K.; Gunasekera, S. P.; Longley, R. E.; Pomponi, S. A. J. Nat. Prod. 2002, 65, 59–61.